Principles of Total Reflection X-Ray Fluorescence Spectroscopy

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Abstract:- the full reflection X-ray visible light (TXRF) technique has been used for the analysis of assorted sorts of solid materials of biological, geological and environmental origin. X-ray visible light spectroscopic analysis (XRF) has reached the stage where it's classified as a mature analytical technique. The theoretical principles area unit well understood. Additionally trendy instrumentation demonstrates enhanced generator and temperature stability. High sensitivity is possible even for light-weight elements and effective matrix correction software package is getable to the modern day spectroscopes. Apart from its continued applications in analysis and development, XRF has become a routine methodology management tool. X-ray Powder optical phenomenon (XRD), on the alternative hand, has with minor exceptions as inside the cement business, principally remained an exploration tool, despite being associate older analytical technique than XRF. XRD has progressed significantly inside the past decade from a in the main qualitative technique for the identification of crystalline materials to a quantitative tool with the advance of further powerful software package packages. This software package has improved instrument management, but collectively quantification and structure determination pattern the Rietveld technique. Consequently, XRD is quickly getting in the strategy management atmosphere.

This paper proposes the standardization of total reflection X-ray visible light (TXRF) analysis for environmental and biological samples. The importance of TXRF in environmental and biological analysis is presently well established and it's endlessly growing. However in these fields there do not appear to be standardized ways in which according by the International commonplace Organization (ISO). For this reason, the international collaboration for the event of recent commonplace procedures and ways in which for TXRF analysis is required. Indeed, a project for "Inter laboratory comparison of TXRF spectroscopic analysis for environmental analysis" was originated inside the frame of the Versailles Project on Advanced Materials and Standards, with the aim of developing a replacement ISO commonplace for TXRF dedicated to environmental and biological analysis.

Total Reflection X-ray visible light (TXRF), a classy variant of Energy Dispersive X-ray visible light (EDXRF), might be a comparatively new technique of material characterization. The geometrical enhancements in TXRF lead to betterment of detection limits by several orders of magnitude compared to it of EDXRF. TXRF are going to be used primarily in three sorts of applications: component analysis, tiny analysis and depth identification.

Key Words: - Total reflection X-ray visible radiation (TXRF), X-ray Powder optical phenomenon (XRD), X-ray visible radiation spectroscopy (XRF), Sample Preparation

1. INTRODUCTION

1.1 X-ray Fluorescence Spectroscopy (XRF)

XRF could be a sensible general purpose analytical tool which will analyze each solid and liquid sample. It's supported the impact of the interaction of x-rays with matter that causes excitation of sure electrons that once quiet to their state; emit x-rays of a characteristic wavelength for that component. This technique, which may be non-destructive, is extremely sensible for distinguishing impurities in metals or characterizing a specific alloy. it's going to even be used for characterizing impurity levels of chemical compounds. Detection limits vary with the character of the sample and therefore the parts present however typically of the order of 0.1% by weight. For solid samples, this system will solely analyze the primary few hundred microns of the surface. For this reason, it's not appropriate for heterogeneous solid mixtures.

1.2 Total Reflection X-ray Fluorescence Spectroscopy (TRXRF)

This technique is analogous to XRF except that the incident x-rays make a really shallow angle to the substrate and don't penetrate it. Instead, characteristic radiation is created by the terribly top layers of the sample as well as any contaminants. To be effective, the substrates ought to be flat and this has crystal rectifier to that being adopted for the identification of semiconductor contaminants on wafers. However, alternative samples could also be examined providing they're sufficiently flat. Detection limits are contingent the weather gift however are within the ppb to ppm vary.

1.3 X-ray diffraction from powders and solids to thin films and nonmaterial's

In materials analysis, the individual has several analytical queries associated with the crystalline constitution of material samples. XRD is that the solely laboratory technique that reveals structural data, like chemical composition, crystal structure, crystallization size, strain, most wellliked orientation and layer thickness. Materials researchers thus use XRD to research a large varies of materials, from powders and solids to skinny films and nonmaterial's.

2. SCIENCE AND INDUSTRY

Many researchers, in industrial similarly as in scientific laboratories, trust diffraction as a tool to develop new materials or to boost production efficiency. Innovations in diffraction closely follow the analysis on new materials, like in semiconductor technologies or pharmaceutical investigations. Industrial analysis is directed towards ever-increasing speed and efficiency of production processes. Totally automatic x-ray diffraction analysis in mining and building materials production sites leads to more cost effective solutions for production management.

3. SOLUTIONS FOR ANALYTICAL QUESTIONS

X-ray diffraction analysis meets several of the analytical desires of a materials individual. In known powders, chemical phases are qualitatively still as quantitatively. Highdiffraction resolution reveals the laver parameters like composition, thickness, roughness and density in semiconductor thin films. Small-angle X-ray scattering and pair distribution function (PDF) analysis facilitate to research the structural properties of nano materials. Stresses and most well-liked orientation may be determined during a big selection of solid objects and built elements.

X-ray fluorescence (XRF) spectrum analysis is an elemental analysis technique with broad application in science and industry. XRF relies on the principle that individual atoms, once excited by an external energy source, emit X-ray photons of a characteristic energy or wavelength. By count the quantity of photons of every energy emitted from a sample, the weather present could also be identified and quantities.

Henry Moseley was perhaps the father of this system, since he, building on W.C. Röentgen's discovery of the high-energy radiation dubbed X-rays, designed an x-ray tube that he accustomed bombard samples with high-energy electrons. Moseley in 1912 discovered a mathematical relationship between the elements's emitted X-ray frequency and its atomic number. In 1925 Coster and Nishina were the first to use primary X-rays rather than electrons to excite a sample. When Glocker and Schreiber were the primary to perform measure of materials victimisation XRF in 1928, detector technology had to catch up so as to form the technique sensible, that didn't begin to happen till the 1940's. The 1950's saw the primary commercially created X-ray spectrometers. In 1970, the metallic element drifted element detector was developed, and this technology continues to be in use these days (Jenkins 1988: 51-53).

Modern XRF instruments present capable of analyzing solid, liquid, and thin-film samples for each major and trace (ppm-level) parts. The analysis is fast and typically sample preparation is nominal or not needed the least bit.

The Archaeometry Laboratory at the letter of the alphabet analysis Reactor is that the home of 2 Bruker Tracer III-V hand-held instruments (one equipped with an automatic sample changer) and an Elva-X bench top instrument. These instruments are used primarily for the provenance research on obsidian artefacts from round the world; however they're conjointly employed in special circumstances for the non-destructive analysis of alternative materials like metals, ceramic paints, and soils.

4. MAIN PRINCIPLE

The main principle of X-Ray fluorescence spectroscopy (XRF) relies on the actual fact, that atoms, once irradiated with X-Rays, radiate secondary X-Rays –the fluorescence radiation.

On this basis XRF-analysis is feasible because:

• The wavelength and energy of the fluorescence radiation is particular for every component.

• The concentration of every component may be calculated victimisation the intensity of the fluorescence radiation.

5. TXRF ANALYSIS USING THE PICOTAX

The regulation of Total reflection X-Ray visible radiation spectrometry as complete within the PicoTAX spectroscope is shown in Figure one.



The X-Ray beam, generated by the molybdenum tube, is mirrored on a Ni/C-multilayer leading to a monochromatic X-Ray beam. This small beam passes the sample holder carrying the sample at a awfully small angle $(0.3 - 0.6^{\circ})$ inflicting total reflection of the beam. The characteristic fluorescence radiation emitted by the sample is detected by an energy dispersive detector (XFlash® detector) and also the intensity is measured by means that of electronic equipment coupled to a multi-channel instrument.

The main distinction with relevance common XRF spectrometers is that the use of monochromatic radiation and also the total reflection optic. Illuminating the sample with a completely mirrored beam reduces the absorption yet because the scattering of the beam within the sample and its matrix.



Resulting advantages are a mostly reduced ground noise, and consequently a lot of higher sensitivities and also the considerably reduction of matrix effects.

One major advantage of TXRF, compared to atomic spectrometry strategies like AAS or ICP-OES, is that the rejection of memory effects.

The technical parameters of the PicoTAX spectrometer are summarized in Table 1.

Table 1. Tech	nical parameters of the PicoTAX spectrometer
Tube	Metal-ceramic (50 W) Air-cooled Mo-Anode
Optics	Ni/C-Monochromator (17,5 keV)
Detector	XFlash® Area: 10 mm ² FWHM: < 160 eV @ MnKα
Size	41,2 x 59,0 x 30,0 cm
Weight	37 kg



6. SAMPLE TYPES AND PREPARATION

An outline of samples varieties, which may be analyzed by means that of TXRF is given in Table one, showing the good variety of applications.

For TXRF analysis all samples should be ready on a sample receptacle that reflects X-Ray radiation. For this purpose the usage of trays with a diameter of 30mm, made from acrylic or crystal is common.

Table 2: Sample (KLOCK	e materials analyzable by ENKÄMPFER, 1997)	TXRF
Liquids	Solids (Anorganic)	Solids (Biogenous)
Water potable, nver, rain, sea and waste water	Soit: mud, sediments, sewage sludge	Plant materiač algae, hay, leaves, braid, moss, needles, roots, wood
Body fluids blood, serum, urine	Suspended particles: aerosols, dusts, flue ash Minerals	Nutrition: fish, (sea-) fruits, meat, mush- rooms, nuts, vege-
	ores, rocks, silicates, silicon	tables
Pure chemicals acids, bases, sol- vents, water	Pigments creams, inks, oil paint, powder	Tissue hair, kidney, liver, lung, nails
	Metals: aluminium, iron, steel	
Oils and crude oil combustibles, crude oil, fat and grease	Thin layers: contaminations, films, foils, layers, precipitates	

Liquids may be perpetrated directly on the sample tray. An amount of many μ l is transferred to the glass disc employing a measuring instrument and afterwards gaseous in desiccators or drying oven (Figure 3).



For solid samples totally different forms of preparation are possible. Powdered samples (suspended matter, soils, minerals, metals, pigments, biogenous solids etc.) may be analyzed directly once preparation of the material on the sample tray. Typically, a number of μg of sample material are transferred, employing a Q-tip or a lint-free tissue.

In a similar manner the direct preparation of single small samples (particles, slivers etc.) is feasible.

Alternatively, powdered solids may be ready as a suspension with volatile solvents like resolving or alcohol. The suspension is then pipette onto the sample tray.

In Figure three the sample preparation is represented for solids that were digestible by microwave digestion.

7. ANALYSIS AND QUANTIFICATION

In general all elements starting from sodium up to uranium (excl. Niobium, molybdenum and Technetium) are often analyzed by the PicoTAX (Figure 4).



PicoTAX spectrometer

Typical TXRF Results Multi-element standard

Element	Conc./(mg/l)	LLD/(mg/l)
Ca	0.991	0.003
Ti	0.997	0.003
٧	0.986	0.002
Cr	0.994	0.002
Mn	1.015	0.002
Fe	0.996	0.001
Co	1.006	0.001
Ni	1.005	0.001
Cu	1	0.001
Zn	0.989	0.001
As	1.019	0
Se	1.1	0
Sr	0.987	0

TXRF analysis is predicated on internal standardization.

Therefore, an element, that isn't present within the sample, should be further for quantification (Figure 3).

The complete method of research and quantification is delineating by the subsequent steps:

• Measure of the entire spectrum.

All detectable elements are measured at the same time.

• Analysis of the measured spectra

All known elements need to be marked for more quantification, which may be done manually or mechanically by the software.

• Spectra de-convolution

On the premise of the chosen elements, the software performs the de-convolution of the spectra.

The net intensities of the element peaks are calculated with relation to corrections of line overlaps, background factors, escape peak correction etc.

Calculation of concentrations

The element concentration is calculated by the straightforward formula:



Where N is that the internet intensity, S the relative sensitivity and c the concentration - each either of the analyse x or the inner customary is, as indicated.

The typical detection limits of the PicoTAX in binary compound solutions are given in Figure five.



8. SUMMARY

The PicoTAX TXRF spectrometer is a versatile instrument for element analysis of various varieties of samples. It's utterly independent of any cooling media and thus applicable for onthe-scene analysis.

Further edges of the PicoTAX are the straightforward standardization routine, the absence of matrix or memory effects and therefore the ability for quick multi-element analysis.

REFERENCES

- Anzelmo, John A. and Lindsay, James R. Journal of Chemical Education, August 1987, 64(8). A181-A185.
- Anzelmo, John A. and Lindsay, James R. Journal of Chemical Education, September 1987, 64(9). A200-A204.
- 3. Buhrke, Victor E.; Jenkins, Ron; Smith, Deane K. A Practical Guide for the Preparation of Specimens for X-Ray Fluorescence and X-Ray Diffraction

Analysis. 1998, John Wiley & Sons, Inc. QD96.X2P73 ISBN 0-471-19458-1.

- 4. Ferrero, J. L. et al. *X-Ray Spectrometry*, 2002, **31**, 441-447.
- Holynska, Barbara et al. Fresenius Journal of Analytical Chemistry, 1998, 362, 294-298.
- Jenkins, Ron, X-Ray Fluorescence Spectrometry. 1988, John Wiley & Sons, Inc. QD96.X2J47 ISBN 0-471-83675-3.
- Jenkins, Ron; Gould, R. W.; Gedcke, Dale. Quantitative X-Ray Spectrometry. Second Edition, 1995, Marcel Dekker, Inc. QD96.X2 J46 ISBN 0-8247-9554-7.
- Jenkins, Ron; Gould, R. W.; Gedcke, Dale. Applied Spectroscopy Reviews, 2000, 35(1,2), 129-150.
- Klockenkämper, Reinhold. Total Reflection X-Ray Fluorescence Analysis. 1997, John Wiley & Sons, Inc. QD96.X2K58 ISBN 0-471-30524-3
- Knoll, G. F. Journal of Radioanalytical and Nuclear Chemistry, 2000, 243(1), 125-131.
- Metz, Uwe; Hoffman, Peter; Weinbruch, Stephan; Ortner, Hugo N. *Mikrochimica Acta*, 1994, **117**, 95-108.
- 12. Potts, Philip J. et al. *Journal of Analytical Atomic Spectrometry*, 2001, **16**, 1217-1237.
- 13. Revenko, Anatoly G. *X-Ray Spectrometry*, 2002, **31**, 264-273.
- 14. Sieber, J. R. X-Ray Spectrometry, 2000, 29, 327-338.
- Skoog, Douglas A.; Holler, F. James; Nieman, Timothy A. Principles of Instrumental Analysis. Fifth Edition, 1998, Thomson Learning, Inc. ISBN 0-03-002078-6.
- 16. Spectro X-Lab Pro Manual, circa 2000. Spectro Analytical Instruments, Inc.
- 17. 16th International Congress of Metrology, 10010 (2013)
- Klockenkämper: Total-Reflection X-Ray Fluorescence Analysis, John Wiley & Sons, 1997.
- 19. VAMAS <u>www.vamas.org</u>
- 20. E. Margui, J.C. Tapias, A. Casas, M. Hidalgo, I.Queralt "Analysis of inlet and outlet industrial wastewater effluents by means of a bench top total reflection X-ray fluoresce
- 21. nce", Chemosphere, Vol.80, pp.263-270, 2010